

# **Test Definition: AMMO**

Ammonium, 24 Hour, Urine

#### Overview

#### **Useful For**

Diagnosis of the cause of acidosis

Diagnosis and treatment of kidney stones

#### **Special Instructions**

• Urine Preservatives-Collection and Transportation for 24-Hour Urine Specimens

#### Method Name

Enzymatic

# NY State Available

Yes

# Specimen

Specimen Type Urine

**Necessary Information** 24-Hour volume (in milliliters) is required.

# **Specimen Required**

#### Supplies:

-Sarstedt 5 mL Aliquot Tube (T914)

-Diazolidinyl Urea (Germall) 5.0 mL (T822)

# Container/Tube: Plastic vial

Specimen Volume: 4 mL

#### **Collection Instructions:**

1. Add 5 mL of diazolidinyl urea (Germall) as preservative at start of collection or refrigerate specimen during and after collection.

2. Collect urine for 24 hours.

3. Aliquot urine into plastic vial.

4. Specimens with pH >8 may indicate bacterial contamination and testing will be cancelled. Do not attempt to adjust pH as it will adversely affect results.

Additional Information: See <u>Urine Preservatives-Collection and Transportation for 24-Hour Urine Specimens</u> for multiple collections.

#### Forms



Ammonium, 24 Hour, Urine

If not ordering electronically, complete, print, and send a <u>Renal Diagnostics Test Request</u> (T830) with the specimen.

## **Urine Preservative Collection Options**

**Note:** The addition of preservative **must occur at the start of collection** or application of temperature controls must occur during and after collection.

| -                                  |           |
|------------------------------------|-----------|
| Ambient                            | No        |
| Refrigerate                        | ОК        |
| Frozen                             | ОК        |
| 50% Acetic Acid                    | ОК        |
| Boric Acid                         | ОК        |
| Diazolidinyl Urea                  | Preferred |
| 6M Hydrochloric Acic               | ОК        |
|                                    |           |
| 6M Nitric Acid                     | No        |
| 6M Nitric Acid<br>Sodium Carbonate | No<br>No  |
|                                    | -         |
| Sodium Carbonate                   | No        |

#### Specimen Minimum Volume

1 mL

## Reject Due To

All specimens will be evaluated at Mayo Clinic Laboratories for test suitability.

#### Specimen Stability Information

| Specimen Type | Temperature              | Time     | Special Container |
|---------------|--------------------------|----------|-------------------|
| Urine         | Refrigerated (preferred) | 14 days  |                   |
|               | Ambient                  | 72 hours |                   |
|               | Frozen                   | 14 days  |                   |

# Clinical & Interpretive

#### **Clinical Information**

The kidney regulates acid excretion and systemic acid base balance. Changing the amount of ammonium in the urine is one important way the kidneys accomplish this task. Thus, measuring the urine ammonium level can provide understanding of the cause of an acid base disturbance in individual patients.(1-3)

The urine ammonium level can also provide a lot of information about the daily acid production in a given patient. Since most of an individual's acid load comes from ingested protein, the urine ammonium is a good indicator of dietary protein intake.

Urine ammonium measurements can be particularly helpful for the diagnosis and treatment of kidney stone patients: -High urine ammonium and low urinary pH suggest ongoing gastrointestinal losses. Such patients are at risk of uric acid

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and calcium oxalate stones.

**1AYO CLINIC** ABORATORIES

-Low urine ammonium and high urine pH suggest renal tubular acidosis. Such patients are at risk of calcium phosphate stones.

-Patients with calcium oxalate and calcium phosphate stones are often treated with citrate to raise the urine citrate (a natural inhibitor of calcium oxalate and calcium phosphate crystal growth). However, citrate is metabolized to bicarbonate (a base), which can increase the urine pH. If the urine pH gets too high, the risk of calcium phosphate stones may have unintentionally been increased. Monitoring the urine ammonium concentration is one way to titrate the citrate dose and avoid this problem. A good starting citrate dose is about one-half of the urine ammonium excretion (in mEq of each). One can monitor the effect of this dose on urine ammonium, citrate, and pH values, and adjust the citrate dose based upon the response. A fall in urine ammonium should indicate whether the current citrate is enough to partially (but not completely) counteract the daily acid load of that given patient.(4)

#### **Reference Values**

15-56 mmol/24 hour

Reference values have not been established for patients <18 years and >77 years of age. Reference values apply to 24 hour collections.

#### Interpretation

If a patient has acidosis and the amount of ammonium in the urine is low, this is suggestive of a renal tubular acidosis.

If the amount of ammonium is high, this suggests that the kidneys are working normally and that there are other losses of bicarbonate in the body. Typically this implies gastrointestinal losses.

#### Cautions

The presence of sulfasalazine, sulfapyridine, or temozolomide may lead to false results.

Ammonium concentrations may be falsely low in samples with a pH above8.0. Consider contamination and/or a urinary tract infection with a urease positive organism (including *Ureaplasma urealyticum*).

#### **Clinical Reference**

1. Peonides A, Levin B, Young W: The renal excretion of hydrogen ions in infants and children. Arch Dis Child. 1965 Feb;40(209):33-39

2. Kamel KS, Briceno LF, Sanchez MI, et al: A new classification for renal defects in net acid excretion. Am J Kidney Dis. 1997 Jan;29(1):136-146

3. Madison LL, Seldin DW: Ammonia excretion and renal enzymatic adaptation in human subjects, as disclosed by administration of precursor amino acids. J Clin Invest .1958 Nov;37(11):1615-1627

4. Coe FL, Evan A, Worcester E: Pathophysiology-based treatment of idiopathic calcium kidney stones. Clin J Am Soc Nephrol. 2011 Aug;6(8):2083-2092

# Performance

#### **Method Description**

Urine samples are diluted 1:100 with clinical laboratory reagent water using a liquid handler and then are analyzed on a Roche cobas.(Package insert: Roche NH3L kit. Roche Diagnostics; V10/2016)



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# PDF Report

No

# Day(s) Performed

Monday through Sunday

# Report Available

Same day/1 to 2 days

## **Specimen Retention Time**

7 days

# **Performing Laboratory Location**

Mayo Clinic Laboratories - Rochester Main Campus

# Fees & Codes

#### Fees

- Authorized users can sign in to <u>Test Prices</u> for detailed fee information.
- Clients without access to Test Prices can contact <u>Customer Service</u> 24 hours a day, seven days a week.
- Prospective clients should contact their account representative. For assistance, contact <u>Customer Service</u>.

#### **Test Classification**

This test has been modified from the manufacturer's instructions. Its performance characteristics were determined by Mayo Clinic in a manner consistent with CLIA requirements. This test has not been cleared or approved by the US Food and Drug Administration.

#### **CPT Code Information**

82140

# LOINC<sup>®</sup> Information

| Test ID | Test Order Name    | Order LOINC <sup>®</sup> Value |
|---------|--------------------|--------------------------------|
| АММО    | Ammonium, 24 Hr, U | 25308-8                        |

| Result ID | Test Result Name    | Result LOINC <sup>®</sup> Value |
|-----------|---------------------|---------------------------------|
| AM24      | Ammonium, 24 Hr, U  | 25308-8                         |
| TM25      | Collection Duration | 13362-9                         |
| VL91      | Urine Volume        | 3167-4                          |